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LOBOCALONE: A NOVEL SECONDARY METABOLITE FROM THE SOFT CORAL *LOBOPHYTUM CALEDONENSE*

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ABSTRACT.—A novel secondary metabolite, lobocalone (**1**), was isolated from the soft coral *Lobophytum caledonense* collected off the South China Sea. Its structure and relative stereochemistry were established mainly by 1D and 2D nmr and NOESY as (1R*,3R*,4R*)-3-(1-methyl)-4-methyl-4-vinylcyclohexyl methyl ketone.

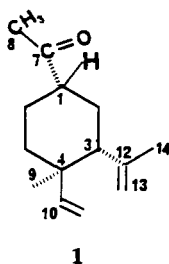
Many cembranoid diterpenes have been isolated from soft corals of the genus *Lobophytum* (1–5). We have previously reported the isolation and structure elucidation of two cembranoid diterpenes from Chinese *Lobophytum* (6,7). Recently, in our research on *Lobophytum caledonense* Tix Dur (Alcyonaceae), we have isolated the new secondary metabolite lobocalone (**1**). We report herein the isolation, structural elucidation, and relative stereochemistry of lobocalone (**1**).

An EtOH extract of dried *L. caledonense* was partitioned between EtOAc and H₂O. Si gel chromatography of the EtOAc extract followed by purification with hplc afforded lobocalone (**1**).

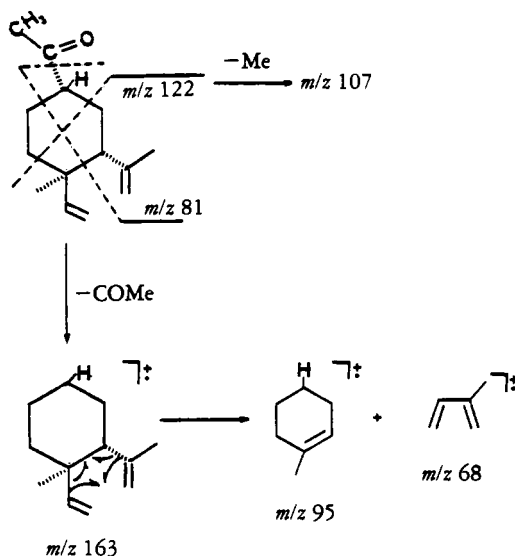
Lobocalone (**1**) was isolated as a colorless oil, $[\alpha]_D^{20} + 18.5^\circ$. Its ir spectrum indicated the presence of a carbonyl group (1709 cm⁻¹) and terminal double bonds (1638, 898 cm⁻¹). A molecular ion at *m/z* 206 and fourteen signals in its ¹³C-nmr DEPT spectrum established the molecular formula as C₁₄H₂₂O with four degrees of unsaturation. The ¹H- and ¹³C-nmr

signals at δ 2.16 and 211.7 and the fragment ion at *m/z* 163 [M–O=CMe]⁺ clearly indicated the presence of a methyl keto group. The 600 MHz ¹H-nmr spectrum showed five vinyl proton signals at δ 4.90, 4.92, 4.60, 4.85, and 5.79. A vinylic methyl signal at δ 1.71 together with four signals of sp² carbons at δ 149.6, 112.6, 146.9, and 110.2 established two terminal double bonds: one is a vinyl group, the other is an isopropenyl group. Based on the data above, **1** must be a monocarbocyclic compound with multifunctionalities.

The ¹H-¹H correlations observed in a 2D ¹H-¹H COSY spectrum enabled us to establish a subunit –CH₂–CH₂–CH–CH₂–CH<. The downfield signal of H-1 (δ 2.39) indicated that the methyl keto group should be linked to C-1. The downfield signal of H-3 (δ 1.98) indicated that C-3 must be connected to one of the double bonds, either the vinyl group or the isopropenyl group. The clear AB splitting of H-10 in its 600 MHz ¹H-nmr spectrum indicated that the vinyl group must be connected with a quaternary carbon (C-4), so C-3 must be connected to the isopropenyl group. Considering the requirement of ring-forming, it was reasonable to assemble the quaternary carbon (C-4) into a six-membered ring with both the vinyl group and the remaining methyl group attached to it. Thus the structure of lobocalone was deduced as **1**. The correlation between



H-3 and the C-14 methyl group observed in the 2D ^1H - ^1H TOCSY spectrum and the ms fragmentation (Scheme 1) also confirmed this deduction.



SCHEME 1. Mass spectral fragmentation of lobocalone [1].

The relative stereochemistry of three chiral carbons in the molecule was determined by a ^1H - ^1H NOESY experiment (Figure 1). A strong nOe between H-1, H-3, and H-5 indicated their diaxial positions; thus both the substituents on C-1 and C-3 must be equatorial. Also a strong nOe between Me-9 and H-2, H-6 was observed, which revealed that Me-9 was axial. Other evidence such as the nOe between H-10 and H-5, H-3 corroborated that Me-9 was axial. As a result, the relative stereochemistry at C-1, C-3, and C-4 was established as $1R^*$, $3R^*$, $4R^*$.

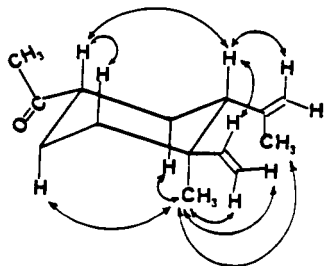


FIGURE 1. Representation of lobocalone [1] showing ^1H - ^1H NOESY correlations.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Their spectrum was measured in CDCl_3 with a Nico-let 5DX-FT spectrometer. ^1H - and ^{13}C -

nmr, ^1H - ^1H COSY, ^1H - ^1H TOCSY, and ^1H - ^1H NOESY spectra were recorded with a Bruker AMX-600 spectrometer. Optical rotations were recorded in MeOH with a Perkin-Elmer 241 polarimeter. Eims was determined at 70 eV on a ZAB-HS spectrometer. Hplc was performed on a Waters 510 liquid chromatograph with a Lambda-Max LC spectrometer, using a μ Porasil Si gel column.

BIOLOGICAL MATERIAL.—The soft coral *L. caledonense* was collected off the South China Sea in May 1990. A voucher specimen was deposited in the Research Center of Organic Natural Products, Zhongshan University, Guangzhou, China.

EXTRACTION AND ISOLATION.—The sun-dried material (930 g) was extracted by EtOH. The extract was evaporated in vacuo to give a dark brown syrup (119.1 g). Then it was partitioned between EtOAc and H_2O . The EtOAc-soluble fraction was evaporated in vacuo to give a brown syrup (43.4 g), which was subjected to flash chromatography (8, 9), using solvents of gradually increasing polarity. The fraction that was eluted with petroleum ether-EtOAc (90:10) was purified by preparative tlc and then by hplc, giving lobocalone [1] (10 mg).

Lobocalone [1].—Colorless oil: $[\alpha]_D^{20} + 18.5^\circ$ ($c=0.13$, MeOH); $\text{ir } \nu_{\text{max}}$ (KBr) 3082, 2988, 2882, 1709, 1638, 1411, 1374, 898 cm^{-1} ; ^1H and

TABLE 1. ^1H - and ^{13}C -nmr Data of Lobocalone [1].^a

Position	$^{13}\text{C}^b$	^1H	^1H - ^1H COSY
1	51.95 (d)	2.39 (m)	H _a -6, H-2
2	29.36 (t)	1.88 (dt, 11, 2.1)	H-3, H-1
3	51.86 (d)	1.98 (dd, 11, 2.9)	H-2
4	39.64 (s)		
5	39.07 (t)	1.49 (dt, 11, 2.4)	H _a -6
6a	23.75 (t)	1.57 (m)	H _b -6, H-5, H-1
6b		1.75 (dd, 11, 2.4)	H _a -6
7	211.7 (s)		
8	28.19 (q)	2.16 (s)	
9	24.74 (q)	1.00 (s)	
10	149.62 (d)	5.79 (dd, 12, 9)	H _a -11, H _b -11
11a	112.84 (t)	4.90 (dd, 9, 2.4)	H-10
11b		4.92 (dd, 12, 2.4)	H-10
12	146.92 (s)		
13a	110.2 (t)	4.61 (d, 1.0)	
13b		4.85 (d, 1.0)	
14	16.46 (q)	1.71 (s)	

^aMeasured at 600 MHz in CDCl₃. Chemical shifts are reported as δ values (ppm) relative to TMS. Multiplicity and J values (Hz) given in parentheses.

^bMultiplicities were determined by DEPT experiment.

^{13}C nmr see Table 1; eims (70 eV) m/z (rel. int.) 206(14), 191(12), 173(7), 163(79), 148(12), 127(31), 121(49), 107(61), 95(100), 81(76), 77(25), 71(35), 68(47), 55(45), 53(34), 51(8).

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